

# Catalytic Engineering Studies of NH<sub>3</sub> Oxidation on Platinum

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Ammonia (NH<sub>3</sub>) is a very important molecule with conflicting roles. On the one hand, NH<sub>3</sub> is a commodity chemical; its manufacture is integral to the production of fertilizers and polyamides. The oxidation of NH<sub>3</sub> on Platinum to NO<sub>x</sub> (x = 1, 2) is one of the oldest commercialized catalytic reactions; it is the basis for the Ostwald Process for the manufacture of nitric acid. On the other hand, NH<sub>3</sub> has emerged as an important reagent in emission control. For the first generation of catalytic converters, NH<sub>3</sub> was an inconvenient byproduct and converter design changes were made to minimize its production. For the modern diesel vehicle, NH<sub>3</sub> is the NO<sub>x</sub> reductant of choice. However, due to NH<sub>3</sub> toxicity, the NO<sub>x</sub> reduction must be accomplished without releasing NH<sub>3</sub> to the environment by Pt-catalyzed NH<sub>3</sub> oxidation. While the Ostwald process relies on high NO<sub>x</sub> yield, the diesel emission control application requires a high N<sub>2</sub> yield.

In this talk we will describe our research in the understanding and design of the structured ammonia slip catalyst (ASC), wherein an NH<sub>3</sub> oxidation function is combined with a selective catalytic reduction (SCR) function. We show that a combination of targeted experiments and modeling enable convergence to an optimal architecture that minimizes the Pt loading. We will also describe advances in the understanding and modeling of NH<sub>3</sub> oxidation kinetics, linking the environmental and chemical manufacturing applications. Finally, we will present a novel core-shell catalyst that comprises a Pt/Al<sub>2</sub>O<sub>3</sub> core and a SCR shell provide for unparalleled ammonia oxidation activity and N<sub>2</sub> selectivity.